

Supplementary data for this paper are available from the IUCr electronic archives (Reference: GD1010). Services for accessing these data are described at the back of the journal.

## References

- Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Grande, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992). *The DIRDIF92 Program System*. Technical Report of the Crystallography Laboratory, University of Nijmegen, The Netherlands.
- Boer, J. L. de & Duisenberg, A. J. M. (1984). *Acta Cryst.* **A40**, C-410.
- Caughlan, C. N. & ul Haque, M. (1967). *Inorg. Chem.* **6**, 1998–2002.
- Enraf–Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Glowiak, T. & Szemik, A. W. (1986). *J. Crystallogr. Spectrosc. Res.* **16**, 79–89.
- Glowiak, T. & Wnek, I. (1985). *Acta Cryst.* **C41**, 324–327.
- Kennard, O., Watson, D. G., Fawcett, J. K., Kerr, K. A. & Coppola, O. C. (1967). *Am. Crystallogr. Assoc. Summer Meet. Abstr.* p. 28.
- Kneeland, D. M., Ariga, K., Lynch, V. M., Huang, C.-Y. & Anslyn, E. V. (1993). *J. Am. Chem. Soc.* **115**, 10042–10055.
- Kral, V., Furuta, H., Schreder, K., Lynch, V. & Sessler, J. L. (1996). *J. Am. Chem. Soc.* **118**, 1595–1607.
- Martinez-Ripoll, M. & Cano, F. H. (1975). *PESOS*. Instituto Rocasolano, CSIC, Madrid, Spain.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Stewart, J. M., Machin, P. A., Dickinson, C. W., Ammon, H. L., Heck, H. & Flack, H. (1976). *The XRAY76 System*. Technical Report TR-446. Computer Science Center, University of Maryland, College Park, Maryland, USA.
- Vetich, G. W. & Caughlan, C. N. (1963). *Acta Cryst.* **16**, A-73.

*Acta Cryst.* (1999). **C55**, 685–687

## 7,11b-Dihydro-9,10-dimethoxy-3,11b-diphenyl[1,2,4]oxadiazolo[5,4-a]-[2,3]benzodiazepin-6(5H)-one

GIUSEPPE BRUNO,<sup>a</sup> ALBA CHIMIRRI,<sup>b</sup> ROSARIA GITTO,<sup>b</sup> FRANCESCO NICOLÓ<sup>a</sup> AND ROSARIO SCOPELLITI<sup>d</sup>

<sup>a</sup>*Dipartimento di Chimica Inorganica, Chimica Analitica e Chimica Fisica, Università di Messina, 98166 Vill. Sant'Agata, Messina, Italy,* and <sup>b</sup>*Dipartimento Farmaco-Chimico, Università di Messina, 98168 Viale Annunziata, Messina, Italy.* E-mail: giuseppe.bruno@unime.it

(Received 28 January 1998; accepted 25 September 1998)

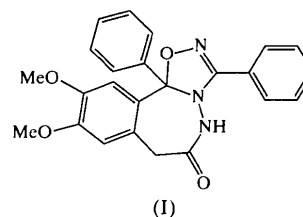
## Abstract

This paper represents one of the few structural reports on 2,3-benzodiazepines. The title compound, C<sub>24</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>, consists of a benzodiazepine moiety and an oxadiazole ring fused together, with the seven-membered ring in a 'boat' conformation. The molecules are linked into pairs by N—H···O hydrogen bonds. The title compound is an interesting antagonist at non-NMDA (*N*-methyl-D-aspartate) receptors.

## Comment

The unusual biological activities shown by benzodiazepine derivatives containing an additional heterocyclic ring fused to different edges of the heptatomic nucleus (Chimirri *et al.*, 1993) have stimulated the exploitation of the chemistry of this class of compounds. Previous reports claim that the biological activity of benzodiazepines can be correlated to the conformational mobility of the seven-membered ring and therefore the fusion of a heterocyclic ring can lead to an enhancement of the ring inversion barrier, thus determining a higher activity and/or specificity of action.

On this basis, in the course of our studies on structure–activity relationships of cyclofunctionalized benzodiazepines, with particular reference to their anti-convulsant effects (De Sarro *et al.*, 1992, 1993), we reported the synthesis of a series of 1,2,4-oxadiazolo[5,4-*a*][2,3]benzodiazepine derivatives (De Sarro *et al.*, 1995), as potential antagonists at non-NMDA receptors, which constitute new interesting targets for anti-epileptic therapy (Chimirri *et al.*, 1999). In this paper, we report the results of the X-ray structure determination of 7,11b-dihydro-9,10-dimethoxy-3,11b-diphenyl[1,2,4]oxadiazolo[5,4-*a*][2,3]benzodiazepin-6(5H)-one, (I), in order to study the geometry of this new class of annelated 2,3-benzodiazepines. We then compare the structure of (I) with those of other known non-NMDA antagonists in order to clarify which structural elements are necessary for anticonvulsant activity.



(I)

There are several reports in the literature describing benzodiazepines, but only a few refer to 1,2- or 2,3-benzodiazepines (Allen & Kennard, 1993). In this paper we report the crystal structure of a 2,3-benzodiazepine which reveals that in the solid state a strong hydrogen bond occurs between molecules related by an inversion centre (see Table 2). Such interaction may be described in terms of the graph set  $R_2^2(8)$  (Bernstein *et al.*, 1995), which is typical for carboxylic acids, amides and their supramolecular association complexes (Walsh *et al.*, 1997).

The title compound consists of three fused rings: the five-membered ring has a twisted envelope conformation [ $\varphi_2 = -130.8(4)^\circ$ ] whereas the seven-membered ring shows the usual 'boat' conformation [ $\varphi_2 = 100.98(9)^\circ$ ,  $\varphi_3 = 159.6(4)^\circ$ ,  $Q = 0.9688(17) \text{ \AA}$ ,  $\theta = 78.37(10)^\circ$ ; Cremer & Pople, 1975]. The disposition of the rings is shown by the torsion angles

C4—C5—C6—C7 = 4.0 (2) and N2—N3—C4—O14 = -118.46 (15)°. The seven-membered ring may be considered as being formed of two parts: the first is coplanar with the dimethoxybenzene ring and the second C1,N2,N3 forms an angle of 81.6 (1)° with the six-membered ring. Geometrical parameters of the seven-membered ring are within the range of values reported for analogous compounds; the single N—N bond length [1.4016 (18) Å] is comparable with the value reported for 1-(3,4-dimethoxyphenyl)-4-methyl-5-ethyl-7,8-dimethoxy-5H-2,3-benzodiazepine hydrobromide (Fogassy *et al.*, 1986) and is slightly smaller than the value found in 6-(*p*-chlorophenyl)-3-(*p*-chlorobenzoyl)-4H-pyrazolo[4,3-*d*][2,3]benzodiazepine (Laskos *et al.*, 1995).

The sum of the valency bond angles around N3 is 346.70 (13)° demonstrating its pyramidalization; the orientation of its lone pair favours the interaction with H14 and H26 (see Table 2). The phenyl substituent at C4 is arranged in such a way to allow an intramolecular interaction between O14 and H18 (see Table 2). The carbonyl double bond and C1—N2 [1.2305 (19) and

1.346 (2) Å, respectively] are indicative of the slight electronic delocalization favoured by hydrogen-bond formation.

The relative disposition of the six- and seven-membered rings is determined by the benzenic bond between C5 and C6, while the disposition of the oxadiazole with respect to the 2,3-diazepine ring is determined by the presence of the phenyl substituent at C4. The distances in the five-membered ring are in good agreement with those reported in the literature, showing the presence of a double bond between C12 and N13 [1.285 (2) Å].

## Experimental

The title compound was synthesized as described in De Sarro *et al.* (1995) and recrystallized from ethanol.

### Crystal data

C<sub>24</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>  
*M<sub>r</sub>* = 415.44  
 Triclinic  
*P* $\bar{1}$   
*a* = 9.308 (2) Å  
*b* = 10.6395 (19) Å  
*c* = 11.225 (2) Å  
 $\alpha$  = 86.959 (16)°  
 $\beta$  = 70.753 (19)°  
 $\gamma$  = 79.375 (18)°  
*V* = 1031.5 (4) Å<sup>3</sup>  
*Z* = 2  
*D<sub>x</sub>* = 1.338 Mg m<sup>-3</sup>  
*D<sub>m</sub>* not measured

Mo *K*α radiation  
 $\lambda$  = 0.71073 Å  
 Cell parameters from 30 reflections  
 $\theta$  = 5.65–14.84°  
 $\mu$  = 0.093 mm<sup>-1</sup>  
*T* = 294 (2) K  
 Irregular  
 0.26 × 0.24 × 0.12 mm  
 Colourless

### Data collection

Siemens *R3m/V* diffractometer  
 $\omega$ -2 $\theta$  scans  
 Absorption correction: none  
 3889 measured reflections  
 3644 independent reflections  
 2167 reflections with *I* >  $\sigma(I)$

*R*<sub>int</sub> = 0.035  
 $\theta_{\max}$  = 25.05°  
*h* = 0 → 11  
*k* = -12 → 12  
*l* = -12 → 13  
 3 standard reflections every 197 reflections  
 intensity decay: 2.48%

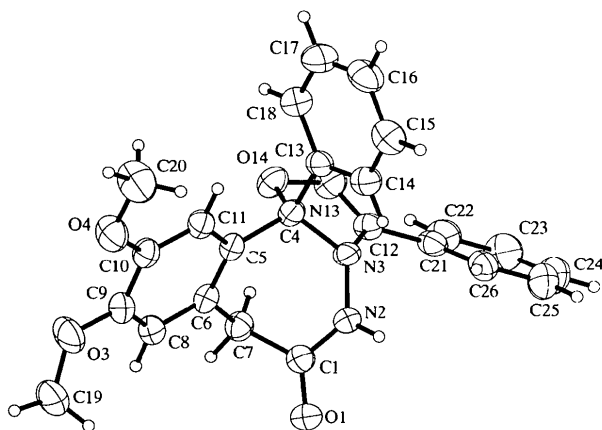


Fig. 1. A view of (I) showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level for non-H atoms.

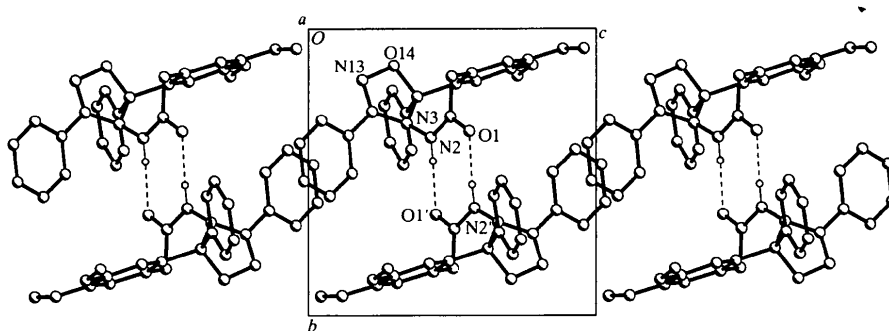


Fig. 2. The molecular packing in (I) viewed down the *a* axis. Only the H atoms involved in hydrogen bonding (indicated as dashed lines) are shown, for clarity.

**Refinement**

Refinement on $F^2$	$\Delta\rho_{\max} = 0.180 \text{ e } \text{\AA}^{-3}$
$R[F^2 > 2\sigma(F^2)] = 0.034$	$\Delta\rho_{\min} = -0.130 \text{ e } \text{\AA}^{-3}$
$wR(F^2) = 0.080$	Extinction correction:
$S = 0.796$	<i>SHELXL97</i> (Sheldrick, 1997)
3644 reflections	Extinction coefficient:
285 parameters	0.0146 (16)
H atoms: see below	Scattering factors from
$w = 1/[\sigma^2(F_o^2) + (0.045P)^2]$	<i>International Tables for</i>
where $P = (F_o^2 + 2F_c^2)/3$	<i>Crystallography</i> (Vol. C)
$(\Delta/\sigma)_{\max} < 0.001$	

Table 1. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

O1—C1	1.2305 (19)	C9—O3	1.356 (2)
C1—N2	1.346 (2)	C10—O4	1.369 (2)
N2—N3	1.4016 (18)	C12=N13	1.285 (2)
N3—C12	1.405 (2)	N13—O14	1.4429 (18)
N3—C4	1.484 (2)	O3—C19	1.414 (2)
C4—O14	1.4492 (19)	O4—C20	1.413 (2)
O1—C1—N2	120.79 (16)	C12—N3—C4	104.02 (13)
O1—C1—C7	122.66 (15)	C12=N13—O14	106.41 (13)
N2—C1—C7	116.17 (15)	N13—O14—C4	106.78 (11)
C1—N2—N3	124.67 (15)	C9—O3—C19	117.87 (15)
N2—N3—C12	118.12 (13)	C10—O4—C20	117.72 (14)
N2—N3—C4	124.56 (13)		

Table 2. Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N2—H2 $\cdots$ O1 <sup>i</sup>	0.859 (17)	1.983 (18)	2.830 (2)	168.5 (17)
C14—H14 $\cdots$ N3	0.93	2.54	2.819 (2)	98
C26—H26 $\cdots$ N3	0.93	2.65	2.944 (2)	99
C18—H18 $\cdots$ O14	0.93	2.50	2.806 (2)	99

Symmetry code: (i)  $1 - x, 1 - y, 1 - z$ .

Reflection intensities were evaluated by profile fitting of a 96-step peak scan in shells of  $2\theta$  (Diamond, 1969) and then corrected for Lorentz-polarization effects. Standard deviations  $\sigma(I)$  were estimated from counting statistics. All non-H atoms were refined anisotropically. H atoms were located in idealized positions (except H2 which was refined isotropically) and allowed to ride on the coordinates of their parent C atoms with a common isotropic displacement parameter ( $U_{\text{iso}} = 0.06 \text{ \AA}^2$ ). All calculations were performed on a MicroVAX 3400 and an AXP DecStation 3000/400.

Data collection: *P3/V Control Software* (Siemens, 1989). Cell refinement: *P3/V Control Software*. Data reduction: *SHELXTL-Plus* (Sheldrick, 1990). Program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994). Program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997). Molecular graphics: *XPW* (Siemens, 1996). Software used to prepare material for publication: *PARST97* (Nardelli, 1995) and *SHELXL97*.

We would like to express our gratitude, for support and aid, to the Italian MURST and to the Centro Interdipartimentale di Servizi per la Diffrazione a Raggi X of the University of Messina.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1353). Services for accessing these data are described at the back of the journal.

**References**

- Allen, F. H. & Kennard, O. (1993). *Chem. Des. Autom. News*, **8**, 31–37.
- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Chimirri, A., De Sarro, G., De Sarro, A., Gitto, R., Grasso, S., Quartarone, S., Zappalá, M., Giusti, P., Libri, V., Constanti, A. & Chapman, A. G. (1999). *J. Med. Chem.* In the press.
- Chimirri, A., Grasso, S., Monforte, A. M., Romeo, G. & Zappalá, M. (1993). *Heterocycles*, **36**, 601–637, 865–890.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- De Sarro, G., Chimirri, A., De Sarro, A., Gitto, R., Grasso, S., Giusti, P. & Chapman, A. G. (1995). *Eur. J. Pharmacol.* **294**, 411–422.
- De Sarro, G., Zappalá, M., Grasso, S., Chimirri, A. & De Sarro, A. (1993). *Gen. Pharmacol.* **24**, 877–884.
- De Sarro, G., Zappalá, M., Grasso, S., Chimirri, A., Spagnolo, C. & De Sarro, A. (1992). *Mol. Neuropharmacol.* **1**, 195–202.
- Diamond, R. (1969). *Acta Cryst.* **A25**, 43–55.
- Fogassy, E., Acs, M., Toth, G., Simon, K., Lang, T., Ladanyi, L. & Parkanyi, L. (1986). *J. Mol. Struct.* **147**, 143–146.
- Laskos, E., Lianis, P. S., Rosios, N. A., Terzis, A. & Raptopoulou, C. P. (1995). *Tetrahedron Lett.* **36**, 5637–5640.
- Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.
- Sheldrick, G. M. (1990). *SHELXTL-Plus*. Release 4.21/V. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXL97. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Siemens (1989). *P3/V Control Software*. Release 4.21. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Siemens (1996). *XPW in SHELXTL*. Version 5.05. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Walsh, P. L., Maverick, E., Chiefari, J. & Lightner, D. A. (1997). *J. Am. Chem. Soc.* **119**, 3802–3806.

*Acta Cryst.* (1999). **C55**, 687–689

### (1*S*,*R*<sub>S</sub>)-6,7-Dimethoxy-*N*-methyl-1-(*p*-tolyl-sulfinylmethyl)-1-trifluoromethyl-1,2,3,4-tetrahydroisoquinoline

PIERFRANCESCO BRAVO, ELEONORA CORRADI, MARCELLO CRUCIANELLI, STEFANO VALDO MEILLE AND MATTEO ZANDA

*Dipartimento di Chimica, Politecnico di Milano, Via Mancinelli 7, 20131 Milano, Italy. E-mail: meille@dept.chem.polimi.it*

(Received 30 September 1998; accepted 30 November 1998)

**Abstract**

The title compound,  $C_{21}H_{24}F_3NO_3S$ , is an intermediate in the synthesis of enantiomerically pure alkaloids; the molecular structure has been determined to establish its stereochemistry. The molecular conformation is largely determined by intramolecular interactions, while the